54. Metabolic Products of Microorganisms

Part 269¹)

5-Phenylpentadienoic-Acid Derivatives from Streptomyces sp.

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Two new phenylpentadienamides isolated from the culture filtrate of *Streptomyces* sp. were assigned the structures 5-(4-aminophenyl)penta-2,4-dienamide (2) and N^2 -[5-(4-aminophenyl)penta-2,4-dienoyl]-L-glutamine (3). In addition, 5-(4-aminophenyl)penta-2,4-dienoic acid (1) has been isolated, and its spectroscopic characteristics are reported for the first time. Compounds 1-3 exist in both the (2*E*,4*E*)- and (2*E*,4*Z*)-configurations. Electrospray and tandem MS, and HPLC/MS proved to be particularly suitable for the characterization of these metabolites.

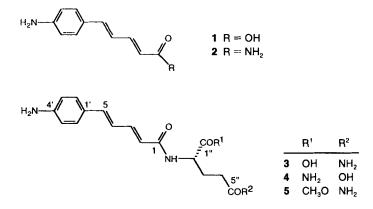
Introduction. – As part of our chemical screening program, we observed that *Streptomyces* sp. Tü 3946 excreted into the culture broth a series of light yellow metabolites. These compounds appeared as orange spots, when TLC plates were sprayed with *Ehrlich* reagent [2]. Here, we report on the isolation and structure elucidation of three of these metabolites. Electrospray MS, which was introduced as a highly sensitive and soft ionization technique for the mass-spectrometric analysis of polar, non-volatile, and thermolabile molecules [3], was used for the characterization of these substances and is described in detail.

Results. – The metabolites were separated from the culture filtrate of *Streptomyces* sp. Tü 3946 by adsorption on *Amberlite XAD 16*. Subsequent fractionation by a combination of gel filtration on *Sephadex LH-20* and reversed-phase chromatography on *C-18* provided compounds **1–3** in pure form (see *Exper. Part*).

HPLC coupled with electrospray MS and diode-array detection revealed that 1–3 were a mixture of two isomers each. Both isomers showed identical fragmentation patterns but differed in their UV-absorption maxima by *ca*. 10 nm (*e.g.* $\lambda_{max} = 305$ and 294 nm for 1). Both isomers were preparatively separated by HPLC. However, on evaporation of the solvent (temperature < 40°), mixtures once again resulted from each one. The isomeric mixtures 1–3 were, therefore, used for the structure elucidation.

Coumpound 1 (mol. wt. 189) was assigned the molecular formula $C_{11}H_{11}NO_2$ by HR-EI-MS. The NMR data (see *Tables 1* and 2) revealed a penta-2,4-dienoic-acid moiety bonded to a *p*-substituted Ph ring. 1D-COSY [4],

¹) Part 268: [1].



HMQC [5], and HMBC [6] experiments allowed complete assignment of ¹H- and ¹³C-NMR signals. Both isomers were shown to differ only in the configuration at the C(4)=C(5) bond. The coupling constant J(4,5) was found to be either 15.4 Hz ((*E*)-configuration) or 11.7 Hz ((*Z*)-configuration; δ (H–C(5)) 6.83 and 6.67 ppm, respectively). On the other hand, J(2,3) (15.1 Hz) was identical in both isomers (δ (H–C(2)) 5.85 and 5.93 ppm, respectively). Corresponding attribution of H–C(2) and H–C(5) was confirmed by a heteronuclear long-range coupling between H–C(2) and C(1) detected in the HMBC spectrum. Thus, compound **1** constitutes a mixture of (2*E*,4*E*)- and (2*E*,4*E*)-isomers of 5-(4-aminophenyl)penta-2,4-dienoic acid. The NMR and EI-MS data are in very good agreement with those reported for (2*E*,4*E*)-5-(4-hydroxyphenyl)penta-2,4-dienoic acid (avenalumic acid) [7].

The molecular formula of compound 2 (mol. wt. 188) was established as $C_{11}H_{12}N_2O$ by HR-EI-MS. The ¹Hand ¹³C-NMR data (see *Tables 1* and 2) were very similar to those of 1. The structure was established to be 5-(4-aminophenyl)penta-2,4-dienamide. As in 1, the C(4)=C(5) bond exists in either the (Z)- or (E)-configuration.

H-Atom	1		2		3	
	$(2E, 4E)^{a}$	(2E,4Z)	(2E,4E)	(2 <i>E</i> ,4 <i>Z</i>)	(2E,4E)	(2 <i>E</i> ,4 <i>Z</i>)
H-C(2)	5.85	5.93	6.00	6.09	6.07	6.16
- /	(d, J = 15.1)	(d, J = 15.1)	(d, J = 15.1)	(d, J = 15.0)	(d, J = 15.0)	(d, J = 15.0)
H-C(3)	7.40	7.84	7.29	7.77	7.26	7.75
	(dd, J = 15.1,	(dd, J = 15.1,	(dd, J = 15.1,	(dd, J = 15.0,	(dd, J = 15.0,	(dd, J = 15.0,
	10.5)	12.1)	10.5)	11.8)	9.7)	11.8)
H-C(4)	6.74	6.19	6.71	6.16	6.68	6.14
	(dd, J = 15.4,	(pseudo-t,	(dd, J = 15.5,	(pseudo-t,	(dd, J = 15.1,	(pseudo-t,
	10.5)	J = 11.6)	10.5)	J = 11.8)	9.7)	J = 11.5)
H-C(5)	6.83	6.67 ^b)	6.79	6.62	6.74	6.57
	(d, J = 15.4)	(d, J = 11.7)	(d, J = 15.5)	(d, J = I1.8)	(d, J = 15.1)	(d, J = 11.3)
H-C(2'/6')	7.27	7.13	7.25	7.15	7.22	7.13
	(d, J = 8.4)	(d, J = 8.4)	(d, J = 8.3)	(d, J = 8.5)	(d, J = 8.5)	(d, J = 8.5)
H-C(3'/5')	6.65	6.69	6.64	6.68	6.64	6.67
	(d, J = 8.4)	(d, J = 8.4)	(d, J = 8.3)	(d, J = 8.5)	(d, J = 8.5)	(d, J = 8.5)
H-C(2")					4.38 (m)	4.38 (m)
H-C(3")					1.95-2.05(m);	1.95-2.05 (m);
					2.15 - 2.25(m)	2.15 - 2.25(m)
H-C(4")					2.4-2.25(m)	2.4-2.25(m)

Table 1. ⁴	H-NMR Data (CD_1OD_1	400.1 MHz)	of Con	npounds 1–3

^{a)} Configuration of the pentadiencyl molety. ^{b)} Obscured; signal observed in a 1D-COSY experiment [4] by selective irradiation of H-C(4) via a 100-ms Eburp-shaped pulse.

C-Atom	1		2		3	
	$(2E, 4E)^{a})$	(2E, 4Z)	(2E,4E)	(2E, 4Z)	(2E, 4E)	(2 <i>E</i> ,4 <i>Z</i>)
C(1)	171.3	171.2	171.8	171.6	168.6	168.4
C(2)	119.4	122.9	121.5	124.8	122.5	125.7 ^b)
C(3)	147.8	143.0	144.2	139.7	142.9	138.5
C(4)	122.8	124.6	123.0	124.8	123.2	125.0 ^b)
C(5)	142.9	139.6	141.9	138.7	141.2	138.0
C(1')	126.9	127.3	127.2	127.5	127.2	127.6
C(2'/6')	129.9	131.8	129.6	131.8	129.5	131.7
C(3'/5')	116.0	115.9	116.0	115.9	116.0	115.8
C(4′)	150.8	149.8	150.5	149.6	150.2	149.3
C(1")					178.4 ^c)	178.4°)
C(2")					55.8	55.8
C(3")					30.2	30.2
C(4″)					33.1	33.1
C(5")					178.4 ^c)	178.4°)

Table 2. ¹³C-NMR Data (CD₃OD, 100.6 MHz) of Compounds 1-3. Assignments based on HMQC spectra.

^a) Configuration of the pentadiencyl moiety. ^b) Values interchangeable although these attributions are preferred. ^c) Broad signal, unresolved.

The UV data of 3 suggested that, as 1 and 2, a phenylpentadienoyl moiety was present. Acidic hydrolysis of 3 (6N HCl) and amino-acid analysis by GLC on a chiral phase [8] revealed the presence of L-glutamic acid. The electrospray-MS of 3 showed pseudomolecular ions at m/z 318 ($[M + H]^+$) and 635 ($[2M + H]^+$). In addition, an intense fragment ion was observed at m/z 172 which was attributed to the acylium ion corresponding to the phenylpentadienoyl moiety of the molecule. The cleavage of the amide bond also occurred by collision-induced dissociation (CID) of the $[M + H]^+$ ion with Ar (see Fig.). The reaction of 3 with CH₂N₂ resulted mainly in a monomethylated product 5, as indicated by an increase of the mol. wt. of 14 u, which demonstrated the presence of an acidic group in the molecule. CID of the $[M + H]^+$ ion of this product proved that the acyl part remained unchanged, since the fragment ion at m/z 172 was still the dominant ion in the daughter-ion mass spectrum. All data obtained so far were in agreement with a (4-aminophenyl)pentadienoyl derivative of an amide of glutamic acid. To localize the amide function, compounds 3 and 4 were prepared by condensation of 1 with glutamine or isoglutamine (4,5-diamino-5-oxopentanoic acid), respectively. Both synthetic compounds had a mol. wt. of 317

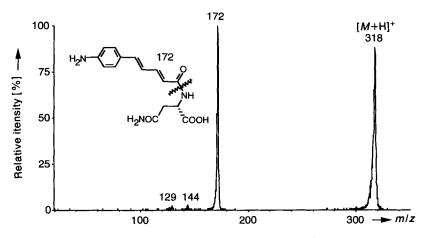


Figure. Daughter-ion mass spectrum of 3 (CID of $[M + H]^+$, m/z 318)

determined by electrospray MS and afforded a dominant fragment ion at m/z 172 in the daughter ions MS of the $[M + H]^+$ peak. While synthetic 3 proved to be identical with the natural compound, 4 clearly differed from the latter on TLC and HPLC analyses. Thus, compound 3 is N^2 -[5-(4-aminophenyl)penta-2,4-dienoyl]-L-glutamine. The C(4)=C(5) bond exists in either the (*E*)- or (*Z*)-configuration.

Discussion. – Compounds 1–3 were not detected in the mycelium. It is not clear, whether (2E, 4E)- and (2E, 4Z)-isomers were both synthesized by the microbial strain, or if isomerization occurred during the fermentation and isolation procedure. Interestingly, avenalumic acid undergoes isomerization on exposure to daylight [7]. The amides 2 and 3 have so far not been described in the literature. The chemical synthesis of 1 has been reported already in the beginning of this century [9], but no spectroscopic data have been available up to now. Compound 1 has been obtained for the first time from a biological source. 5-Phenylpenta-2,4-dienoic-acid derivatives are rather unusual natural products. 5-Phenylpenta-2,4-dienoic acid itself has been detected in the bud exudate of *Populus* species [10] [11]. Some *p*-hydroxy derivatives have also been found in plants: psilotic acid has been isolated from *Psilotum nudum* [12], while several avenalumic acid conjugates were obtained from oat groats and hulls [7]. On the other hand, there is no report about the production of phenyl-pentadienoic-acid derivatives by microorganisms. Compounds 1–3 proved to be inactive against Gram-positive and Gram-negative bacteria in our testing systems.

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Experimental Part

General. H-L-Gin-O'Bu·HCl and H-L-Glu(O'Bu)-NH₂·HCl were purchased from *Bachem Biochemica*. Prep. low-pressure liquid chromatography (LPLC): Lobar C-18 column (40–63 µm; i.d. 2.5 × 27 cm; Merck) equipped with a FR-30 HPLC pump (Knauer). Prep. HPLC: Nucleosil 100 C-18 column (10 µm, i.d. 250 × 16 mm) equipped with a precolumn (i.d. 30 × 16 mm), 2 Sepapress HPP 200/100 high-performance pumps (Kronwald) and a Sepacon GCU-311 gradient control unit (Kronwald). TLC: silica-gel-precoated Al sheet (Macherey-Nagel), BuOH/EtOH/ H₂O 6:2:2 (eluent 1). GLC: GC-Sichromat 1 (Siemens) equipped with a N-selective detector and an Autoderivat 100 derivatization system (CAT, Tübingen, Germany). M.p.: given for the mixture of stereoisomers and are uncorrected. [α]_D: Perkin-Elmer 241 polarimeter; 0.1-dm cell. IR: Perkin-Elmer 281 B infrared spectrophotometer. NMR: High-resolution spectra were obtained on a Bruker AMX 400 spectrometer interfaced to a X32 computer and equipped with an inverse triple resonance probe; 50 mm soln. in CD₃OD were used; chemical shifts were referenced to the solvent peak (δ (H) 3.30 ppm; δ (C) 49.0 ppm); assignments by the use of 1D-COSY [4], heteronuclear multiple quantum coherence (HMQC) [5] and heteronuclear multiple bond correlation (HMBC) [6] experiments.

Microbial Strain. Streptomyces sp. Tü 3946 was isolated from a soil sample collected in 1990 in New Zealand. The strain builds a gray spore mass on yeast extract agar. The spore surface is spiny. Melanin is not produced on peptone/yeast extract/iron agar.

Production and Isolation. The strain Tü 3946 was cultivated in 500-ml *Erlenmeyer* flasks with four baffles containing each 150 ml of a medium consisting of decreased soybean meal (2%), starch (1%), glycerol (1%), and NaBr (0.03%; pH 7.75). Each flask was inoculated with 5 ml of a 36-h old preculture (medium: soybean meal (2%), mannitol (2%), pH 7.5). The fermentation was carried out on a rotary shaker for 4 d at 27°. After filtration of the culture broth (5 l), the culture filtrate was passed through a *Amberlite XAD 16* column (500 ml), and the

compounds were then eluted with MeOH (2 1). After evaporation, a 8.6-g portion of the residue (9.1 g) was fractionated on a *Sephadex LH-20* column (i.d. 61×5 cm) with MeOH/H₂O 1:1. Fourteen fractions were collected (I–XIV). Compound 1 was purified from *Fraction VII* (74 mg) by LPLC on *RP-18* with MeOH/H₂O 4:6; compounds 2 and 3 were obtained following the same procedure from *Fractions XIII* (82 mg) and *V* (930 mg), resp., with MeOH/H₂O 2:8 and 3:7, resp. Final purification of compounds 1 (3 mg), 2 (12 mg), and 3 (117 mg) was achieved by prep. HPLC on *RP-18* with following eluents (flow rate 20 ml/min): *I*. MeOH/0.05% aq. TFA 1:9→1:0 in 30 min; *2*. MeCN/H₂O 0:1→7:3 in 10 min; *3*. MeOH/H₂O 1:9→1:1 in 15 min.

Mass Spectrometry. EI-MS: TSQ 70 spectrometer (Finnigan MAT). HR-EI-MS: Modified MAT 711A spectrometer (AMD INTECTRA). Electrospray MS and tandem MS: Measurements were performed on a Sciex API III triple-quadrupole mass spectrometer with mass range of 2400 Da equipped with a nebulizer-assisted electrospray ('ion spray') ion source (Sciex, Thornhill, Ontario, Canada). The mass spectrometer was operated in positive-ion mode under conditions of unit mass resolution for all determinations. The accuracy of mass determination was \pm 0.1 u. Profile spectra were obtained by acquiring data points every 0.1 Da. Electrospray voltage was +4.9 kV, orifice voltage was +80 V. Collision-induced dissociation experiments (CID, daughter-ion scans) were performed using Ar at a target gas thickness of ca. 5×10^{14} atoms cm⁻² with collision energies of ca. 40 eV. Daughter ions were obtained by electrospray MS and tandem MS (MS/MS) either directly or on-line in combination with a narrow-bore HPLC-system (Applied Biosystem 140A). For direct injection, the soln. was introduced into the electrospray source at a constant flow rate of 5μ /min with a metical syringe infusion pump (Harvard Apparatus, modell 22, Southnatick, USA) in combination with a microliter syringe (100 μ l, Hamilton, # 1710, USA).

HPLC/MS. Analyses were performed on a *Nucleosil C-18* narrow-bore column (5 μ m; 100 × 2 mm) equipped with a precolumn (10 × 2 mm). The column was connected with the electrospray interface *via* a fused capillary (length 30 cm; 100 μ m i.d.). A gradient of aq. MeCN, containing TFA (0.1%), was used as eluent (MeCN/H₂O 0:1 \rightarrow 1:1 in 10 min, flow-rate 200 μ l/min; 40 μ l/min into MS); detection by electrospray MS.

HPLC/UV. HP 1090M Liquid chromatograph equipped with a diode array detection system and a work station (*Hewlett-Packard*). Analyses were performed on a *Nucleosil C-18* column (5 μ m; i.d. 125 × 4.6 mm) equipped with a precolumn (i.d. 20 × 4.6 mm). A gradient of aq. MeCN was used as eluent (MeCN/0.1% aq. H₃PO₄ 0:1 \rightarrow 1:1 in 10 min, flow-rate 2 ml/min).

Amino-Acid Analysis. Compound 3 (0.1 mg) was dissolved in 6N HCl and hydrolyzed for 18 h at 110°. Derivatization was performed directly before the analysis by successive treatment with 3N HCl/PrOH (110°, 30 min) and (CF₃CO)₂O/CF₃COOEt (140°, 10 min). After evaporation of the reagents, the sample was dissolved in toluene/Me₂CO 3:1 and analyzed by GLC on a *Chirasil-Val* glass capillary column (20 m \times 0.3 mm) [8]; the oven temp. programme was 80° for 3 min, then 80° to 190° at 4° min⁻¹. L-Glutamic acid was identified by comparison with a standard.

5-(4-Aminophenyl)penta-2,4-dienoic acid (1, $C_{11}H_{11}NO_2$). Orange-to-brown solid. M.p. 178°. TLC (SiO₂, eluent 1): R_f 0.56. IR (KBr): 3360, 3200, 1675, 1590, 1510, 1295, 1265, 995. HPLC/UV ((2*E*,4*E*)-isomer; r.t., 4.5 min): 305, 230. HPLC/UV ((2*E*,4*Z*)-isomer; r.t., 4.9 min): 294, 224. Electrospray-MS: 190 ([*M* + H]⁺). EI-MS (70 eV): 189 (19), 144 (100), 127 (18), 155 (11). HR-EI-MS ([*M*]⁺): calc. 189.0800, found 189.0795.

5-(4-Aminophenyl)penta-2,4-dienamide (2, $C_{11}H_{12}N_2O$). Yellow solid. M.p. 205°. TLC (SiO₂, eluent 1): R_f 0.63. IR (KBr): 3320, 3200, 1650, 1580, 1510, 1385, 1290, 990. HPLC/UV ((2*E*,4*E*)-isomer; r.t., 3.4 min): 303, 228. HPLC/UV ((2*E*,4*Z*)-isomer; r.t. 3,7 min): 292, 224. Electrospray-MS: 189 ([M + H]⁺). EI-MS (70 eV): 188 (5), 144 (100), 143 (60), 127 (30), 115 (24). HR-EI-MS ([M]⁺): calc. 188.0949, found 188.0944.

 N^2 -[5-(4-Aminophenyl)penta-2,4-dienoyl]-L-glutamine (3, C₁₆H₁₉N₃O₄). Yellow-to-orange solid. M.p. 250° (dec.). [α]_D³⁰ = -4.9 (c = 0.205, DMSO). TLC (SiO₃, eluent 1): R_f 0.36. IR (KBr): 3350, 1650, 1585, 1505, 1395, 1285, 995. HPLC/UV ((2*E*,4*E*)-isomer; r.t., 3.5 min): 306, 230. HPLC/UV ((2*E*,4*Z*)-isomer; r.t., 3.7 min): 295, 225. Electrospray-MS: 318 ([M + H]⁺). Tandem-MS (electrospray-MS, CID of [M + H]⁺): 318, 172.

N²-[5-(4-Aminophenyl)penta-2,4-dienoyl]-L-glutamine Methyl Ester (5, C₁₇H₂₁N₃O₄). A soln. (1 ml) of CH₂N₂ in Et₂O was added to a soln. of 3 (17 mg) in BuOH/MeOH 2:1 (1.5 ml). After stirring for 15 h at r.t., the solvent was evaporated and the crude product purified by gel filtration on Sephadex LH-20 with MeOH/H₂O 1:1. Final purification by HPLC on *RP-18* with MeCN/H₂O (2:8→8:2 in 30 min) provided 3 mg of 4. Yellow solid. ¹H-NMR (CD₃OD, (2*E*,4*E*)-isomer): 7.30 (*dd*, J = 15.0, 10.4, H–C(3)); 7.25 (*d*, J = 8.6, H–C(2'), H–C(6')); 6.80 (*d*, J = 15.5, H–C(5)); 6.72 (*dd*, J = 15.5, 10.4, H–C(4)); 6.65 (*d*, J = 8.4, H–C(3'), H–C(5')); 6.05 (*d*, J = 15.0, 10.4, H–C(3')). Electrospray-MS: 332 ([M + H]⁺). Tandem-MS (electrospray-MS, CID of [M + H]⁺): 332, 172.

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Preparation of **3** *and* **4** *from* **1**. *Compound* **3**. *N*,*N*'-diisopropylcarbodiimid (8.3 µl, 53 µmol), 1-hydroxy-1*H*-benzotriazol (7.2 mg, 53 µmol), Et(i-Pr)₂N (18 µl, 105 µmol), and 4-(dimethylamino)pyridin (0.7 mg, 5.7 µmol) were dissolved in 1 ml of DMF. This soln. (0.1 ml) was added to a soln. of **1** (1 mg, 5.3 µmol) and H-t-Gin-O(*t*-Bu)·HCl (1.4 mg, 5.8 µmol) in DMF (0.1 ml). The mixture was kept at 60° for 15 h. After evaporation of the solvent, the dry residue was treated with 0.4 ml of TFA for 30 min. Purification by HPLC on *RP-18* with MeCN/0.1% aq. TFA (0:1→2:8 in 10 min) provided 0.6 mg of **3** identical (DC, HPLC, HPLC/MS and tandem-MS) with the natural compound. *Compound* **4**. N²-[*5*-(*4*-*Aminophenyl*)*penta-2.4-dienoyl*]-1-*isoglutamine* (**4**) was prepared from **1** (1 mg) and H-t-Giu(O(*t*-Bu))-NH₂·HCl (1.4 mg) following the same procedure. Purification as described for **3** provided 0.8 mg of **4**. TLC (SiO₂, eluent *1*): *R*_f 0.47. Electrospray-MS: 318 ([*M* + H]⁺). Tandem-MS (electrospray-MS, CID of [*M* + H]⁺): 318,172.

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